A Library Synthesis of Pyrazoles by Azomethine Imine Cycloaddition to the Polymer-supported Vinylsulfone

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1,3-Dipolar cycloaddition of azomethine imines to the polymer-supported vinylsulfone was achieved. Pyrazole derivatives bearing various aryl groups were synthesized regioselectively.

In recent years, methods for the combinatorial synthesis of small-molecule libraries for the development of useful pharmaceuticals remain a formidable challenge. The application of solid-phase organic synthesis (SPOS) in the preparation of small compound libraries is of considerable significance in the discovery and development of new drug compounds.¹ The recent success of the pyrazole cyclooxygenase-2 (COX-2) inhibitor has further highlighted the importance of these heterocycles in medicinal chemistry. 2^{-4} Therefore, they are interesting targets in the development of new drug leads using solid-phase combinatorial chemistry.⁵ 1,3-Dipolar cycloaddition is one of the useful synthetic methods for diversed heterocyclic compounds. In the past few years, a considerable number of solid-phase synthesis of heterocycles utilizing 1,3-dipolar cycloaddition has been reported.⁶ Recently, we have demonstrated a practical synthetic route to a constrained β -strand mimetic template via the regioselective 1,3-dipolar cycloaddition of azomethine imines with β -(p-toluenesulfonyl)acrylate.⁷ We applied this method to the combinatorial synthesis of a β -strand mimetic templates library using polymer-supported β -(p-toluenesulfonyl) acrylate.⁸ Herein, we report a regioselective 1,3-dipolar cycloaddition of azomethine imines to polymer-supported β -(p-toluenesulfonyl)acrylate and subsequent β -elimination of the toluenesulfonyl group in the synthesis of pyrazole derivatives.⁹

Scheme 1 shows the general outline of our approach. Fmoc-glycine 2 ($\mathbb{R}^1 = \mathbb{H}$) was attached to the Rink-amino resin 1 in the usual manner.¹⁰ After removal of the Fmoc group (20%) piperidine/ CH_2Cl_2), coupling of (E) -3-(p-toluenesulfonyl)acrylic acid (4) to polymer-supported amide (DIC/HOBt/DIEA) provided 5. Treatment of the resin 5 with excess (10 equiv.) amounts of the azomethine imine, generated in situ treatment of hydrazonyl chloride 6 ($Ar^2 = 4-MeC_6H_4$, $Ar^3 = Ph$) with base. 1,3-Dipolar cycloaddition on polymer-support smoothly proceeded under the mild conditions (r.t., 8 h) to provide the corresponding pyrazoline 7. Subsequently, elimination of a ptoluenesulfonyl group with DBU (10 equiv.), followed by acid cleavage from the polymer-support $(25\% \text{ TFA}/\text{CH}_2\text{Cl}_2)$ afforded the desired pyrazoles 8. The two regioisomeric products 8a and 8b were detected by 1 H NMR. The major isomer was 8a with 84% selectivity in 98% purity (Table 1, Entry 1). It should be noted that the p-toluenesulfonyl group plays an important role, not only as an electron-withdrawing group activating the

Scheme 1. Solid-phase synthesis of pyrazoles via 1,3-dipolar cycloaddition with polymer-supported vinylsulfones.

dipolarophile and inducing regioselectivity but also as a leaving group allowing aromatization of the pyrazoline.¹¹

We next investigated the synthetic diversity of pyrazoles utilizing this method by the combination of amino acids $(R¹)$ and phenylhydrazines $(Ar^2$ and Ar^3). The results are shown in Table 1. The validation of the Ar^2 group afforded the corresponding pyrazoles in good purities with 76–89% regioselectivity, except 2-furyl and 1-naphthyl groups decreased purities and regioselectivity (Entries 1–11). The substituents of the Ar^3 group did not affect product purities and regioselectivity (Entries 12– 18). Both the Ar^2 and Ar^3 were modified (Entries 19–22). However, a bulky substituent of the $R¹$ group considerably decreased regioselectivity (Entry 25).

In summary, we have demonstrated versatile and efficient solid-phase synthesis of pyrazole derivatives in good purities. The polymer-supported β -(p-toluenesulfonyl)acrylate can be utilized to the synthesis of various heterocycles utilizing other 1,3-dipolar cycloaddition.

Chemistry Letters Vol.34, No.3 (2005) 439

Table 1. Synthesis of pyrazole derivatives by 1,3-dipolar cycloaddition on polymer-support

Entry	R ¹	Ar^2	Ar^3	$8a:8b^a$	Purity ^b /%	δ /ppm of Ha/Hb	Rt/min of $8a^c$	MS ^d
1	H	$4-MeC6H4$	Ph	84:16	98 $(83)^e$	7.75/8.69	7.1	335.1
\overline{c}	H	4-MeOC ₆ H ₄	Ph	78:22	96	7.79/8.68	6.5	351.1
3	$\boldsymbol{\mathrm{H}}$	4-biphenyl	Ph	76:24	96	7.96/8.74	8.4	397.2
$\overline{4}$	H	1-naphthyl	Ph	70:30	75	7.26/8.88	7.5	371.1
5	H	$4-NO2-C6H4$	Ph	87:13	84	7.43/8.79	7.0	366.1
6	H	4 -FC $_6$ H ₄	Ph	84:16	97	7.26/8.72	6.9	339.1
7	H	$4 - CF_3 - C_6H_4$	Ph	77:23	94	7.38/8.77	8.0	389.1
8	$\boldsymbol{\mathrm{H}}$	$4-BrC_6H_4$	Ph	85:15	96	7.28/8.72	7.7	399.0
9	H	2-furyl	Ph	76:24	63	7.17/8.72	5.5	311.1
10	H	cyclohexyl	Ph	89:11	92	6.78/8.59	7.1	327.2
11	H	Ph	Ph	85:15	98 $(85)^e$	7.29/8.71	6.6	321.1
12	H	Ph	$4-MeC_6H_4$	86:14	97	7.27/8.65	7.1	335.2
13	H	Ph	4 -CN-C ₆ H ₄	82:18	81	7.34/8.85	6.7	346.1
14	H	Ph	$4-MeOC6H4$	88:12	94	7.26/8.58	6.7	351.1
15	H	Ph	4- <i>i</i> -Pr C_6H_4	82:18	99	7.27/8.66	8.3	385.2^{f}
16	H	Ph	4 -FC $_6$ H ₄	90:10	98	7.30/8.67	6.8	339.1
17	H	Ph	$4-CIC6H4$	82:18	98	7.30/8.71	7.5	335.1
18	H	Ph	$4-BrC_6H_4$	87:13	98	7.30/8.72	7.6	399.0
19	H	$4-MeC6H4$	$4-MeOC6H4$	85:15	96	7.18/8.62	7.0	365.2
20	H	$4-MeC_6H_4$	$4-NO_2-C_6H_4$	74:26	85	7.41/8.73	7.5	380.1
21	H	4 -CN-C ₆ H ₄	$4-MeOC6H4$	77:23	79	7.26/8.82	6.7	376.1
22	H	$4-MeOC6H4$	$4-NO_2-C_6H_4$	87:13	95	7.40/8.66	7.1	396.1
23	Me	Ph	Ph	79:21	92	7.30/8.72	6.9	335.2
24	$CH2C6H5$	Ph	Ph	82:18	88	7.13/8.56	8.5	433.2^{f}
25	i -Bu	Ph	Ph	68:32	91	7.26/8.70	8.3	377.2

^aThe ratio of regioisomers was determined by ¹H NMR (270 MHz). ^bThe purity of the crude product (8a + 8b) after cleavage was determined by reversed-phase HPLC with peak areas detected by UV (254 nm). ^cReversed-phase HPLC analysis was performed with a Hewlett Packard 1100 series equipped with Inertsil-ODS-3 C18 column (GL Science Inc., 4.6×75 mm) and a binary linear gradient (solvent A = water/0.1% formic acid, solvent B = acetonitrile/0.1% formic acid, $10\%B-90\%B$ in 10 min). $\frac{d}{d}$ Electrospray-TOF Mass data were recorded as $[M + H]^+$ on an Applied Biosystems Mariner. ^eIsolated yield based on the loading amount of the resin. ${}^f[M + Na]^+$ was observed.

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